## Pd(0)-Catalyzed Insertion-Cyclization Reaction of 2,3-Allenols with Aryl or Alkenyl Halides. **Diastereoselective Synthesis of Highly Optically** Active trans-2,3-Disubstituted Vinylic Oxiranes

Shengming Ma\* and Shimin Zhao

Laboratory of Organometallic Chemistry Shanghai Institute of Organic Chemistry Chinese Academy of Sciences, 354 Fenglin Lu Shanghai 200032, P. R. China

## Received February 16, 1999

Allenes are a class of compounds with unique reactivities due to the existence of the two orthogonal  $\pi$ -bonds, and they have been found to be very useful intermediates in organic synthesis.<sup>1-3</sup> Among our efforts in the area of allene chemistry,<sup>3</sup> we developed an efficient one-step route to butenolides with potential biological activities via the Pd(0)-catalyzed cyclization reaction of 1,2-allenic carboxylic acids with aryl or 1-alkenyl halides.3c In this paper we give our results on the corresponding reaction of 2,3-allenols (Scheme 1).

The groups of Olsson<sup>4</sup> and Marshall<sup>5</sup> have shown that the Ag<sup>+</sup>catalyzed cyclization reaction of 2,3-allenols affords 2,5-dihydrofuran derivatives, while Tsuji and co-workers<sup>6</sup> reported that the Pd(0)-catalyzed reaction of 2,3-dienols with aryl or alkenyl halides produced  $\beta$ -aryl- or alkenyl- $\beta$ -methyl- $\alpha$ , $\beta$ -unsaturated enones via an insertion  $-\beta$ -elimination mechanism (pathway a, Scheme 1). We are interested in the possibility of the Pd(0)catalyzed reaction of organohalides and 2,3-allenols to form either 2,4-disubstituted-2,5-dihydrofurans (pathway b, Scheme 1) or 2,3disubstituted vinylic oxiranes (pathway c, Scheme 1).

In our first attempt, we synthesized 1,2-octadien-4-ol<sup>7</sup> and studied its Pd(0)-catalyzed cyclization reaction with PhI under various reaction conditions (Scheme 2). Luckily and surprisingly, we observed in our first try the formation of the three-membered ring, i.e., 2-(1'-phenylethenyl)-3-(n-butyl)oxirane 3a, under the conditions for the corresponding cyclization reaction of 1,2-allenyl carboxylic acids (Scheme 2).3c The formation of the 2,5dihydrofuran derivative was not observed, and the Ag<sup>+</sup> salt was not required (compare entries 1 with 2 and 3 with 4, Scheme 2). After we screened several different reaction conditions, the stereoselectivity of formation of the two diastereoisomers was increased to 30:1 by using DMF as the solvent (compare entries 1-4 with 5, Scheme 2). The relative configuration of the major isomer was established to be trans, as determined from the NOESY spectra of 3e (entry 5, Table 1).

Using conditions A (entry 5, Scheme 2) as the standard reaction conditions, we studied the Pd(0)-catalyzed highly diastereoselec-

(1) Schuster, H. F.; Coppola, G. M. Allenes in Organic Synthesis; John Wiley & Sons: New York, 1988. The Chemistry of Ketenes, Allenes, and Related Compounds; Patai, S., Ed.; John Wiley & Sons: New York, 1980; Part 1.

(2) Walkup, R. D.; Kim, S. W.; Wagy, S. D. J. Org. Chem. **1993**, 58, 6486. Fox, D. N. A.; Lathbury, D.; Mahon, M. F.; Malloy, K. C.; Gallagher, T. J. Am. Chem. Soc. **1991**, 113, 2652. Kimura, M.; Fugami, K.; Tanaka, S.; Tamaru, Y. J. Org. Chem. 1992, 57, 6377. Yamamoto, Y. Y.; Al-Masum, M.; Asao, N. J. Am. Chem. Soc. 1994, 116, 6019. Trost, B.; Gerusz, V. J. Am. Chem. Soc. 1995, 117, 5156.

(3) For some of our recent results in allenes, see: (a) Ma, S.; Shi, Z.; Li, L. J. Org. Chem. **1998**, 63, 4522. (b) Ma, S.; Wei, Q. J. Org. Chem. **1999**, 64, 1026. (c) Ma, S.; Shi, Z. J. Org. Chem. **1998**, 63, 6387. (4) Olsson, L.-I.; Claesson, A. Synthesis **1979**, 743.

(5) Marshall, J. A.; Pinney, K. G. J. Org. Chem. 1993, 58, 7180. Marshall,
J. A.; Yu, R. H.; Perkins, J. F. J. Org. Chem. 1995, 60, 5550.
(6) Shimizu, I.; Sugiura, T.; Tsuji, J. J. Org. Chem. 1985, 50, 537.
(7) Olsson, L.-I.; Claesson, A.; Bogentoft, C. Acta Chem. Scand 1974, 28,

765

Scheme 1



Scheme 2<sup>a</sup>

—•— HC 1a	5 mol% Pd(PPh -n-C <sub>4</sub> H <sub>9</sub> PhI 2a	H3)4 +	n-C <sub>4</sub> H <sub>9</sub>
		3a	3a'
entry	conditions	Yield (%) <sup>a</sup> of <b>3a +3a'</b>	3a/3a' <sup>a</sup>
1	K <sub>2</sub> CO <sub>3</sub> , Ag <sub>2</sub> CO <sub>3</sub> ,	67 <sup>b</sup>	18:1
	CH <sub>3</sub> CN, reflux, 18 h		
2	K <sub>2</sub> CO <sub>3</sub> ,CH <sub>3</sub> CN,	62	14:1
	reflux, 18 h		
3	K <sub>2</sub> CO <sub>3</sub> , Ag <sub>2</sub> CO <sub>3</sub> ,	33°	15:1
	CH <sub>3</sub> CN, 65 °C, 18 h		
4.	K <sub>2</sub> CO <sub>3</sub> , CH <sub>3</sub> CN,	$32^d$	20:1
	65°C, 18 h		
5 <sup>e</sup>	K <sub>2</sub> CO <sub>3</sub> , DMF,	71 <sup>b</sup>	30:1
	55 °C, 14 h		

<sup>a</sup> Determined from 300 MHz <sup>1</sup>H NMR spectra of the crude product before separation using CH<sub>2</sub>Br<sub>2</sub> as the internal standard. <sup>b</sup> Isolated yield. <sup>c</sup> 23% of **1a** was recovered. <sup>d</sup> 18% of **1a** was recovered. <sup>e</sup> This set of reaction conditions is defined as conditions A.

tive epoxide formation reaction carefully, and some of our typical examples were listed in Table 1. The following points should be mentioned:

(1) In most cases, the diastereoselectivities of this reaction are fairly high.

(2) Both aryl and 1-alkenyl halides could afford the corresponding products in moderate to good yields (for the reaction with 1-alkenyl iodides, see entries 4 and 11, Table 1).

(3) For the reaction of *p*-bromoiodobenzene, only the carboniodine bond reacted, while the carbon-bromine bond remained intact (entries 10 and 15, Table 1).

Optically active epoxides are considered as important synthetic intermediates.8 Jung et al. studied the rearrangement of vinylic

<sup>(8)</sup> Maruoka, K.; Hasegawa, M.; Yamamoto, H. J. Am. Chem. Soc. 1986, 108, 3827. Maruoka, K.; Ooi, T.; M.; Yamamoto, H. J. Am. Chem. Soc. 1989, 111, 6431. Jung, M. E.; D'Amico, D. C. J. Am. Chem. Soc. 1993, 115, 12208.

 Table 1.
 Pd(0)-Catalyzed Insertion—Cyclization Reaction of 2,3-Allenol and Aryl or Alkenyl Halides

	<b>_•</b> =>		Conc	ditions /	=	₹ <sup>2</sup>	
	НÓ 1 ्	2				V R	1
entry	<b>1</b> , R <sup>1</sup>	<b>2</b> , R <sup>2</sup>	temp (°C)	time (h)	product	yield (%)	dr <sup>a</sup>
1	$n-C_4H_9$	Ph	55	14	3a	71	96/4
2	n-C <sub>4</sub> H <sub>9</sub>	p-MeO-C <sub>6</sub> H <sub>4</sub>	52	16	3b	71	97/3
3	$n-C_4H_9$	p-Me-C <sub>6</sub> H <sub>4</sub>	53	16	3c	70	97/3
4	$n-C_4H_9$	E-2-Ph-ethenyl	54	24	3d	65	92/8 <sup>1</sup>
5	n-C <sub>4</sub> H <sub>9</sub>	naphthyl	64	48	3e	76	96/4
6	Ph	<i>p</i> -Me-C <sub>6</sub> H <sub>4</sub>	52	16	3f	55	95/5
7	Ph	p-MeO-C <sub>6</sub> H <sub>4</sub>	53	16	3g	66	92/8 <sup>i</sup>
8	$n - C_7 H_{15}$	<i>p</i> -Me-C <sub>6</sub> H <sub>4</sub>	54	16	3h	88	97/3
9	$n - C_7 H_{15}$	<i>p</i> -MeO-C <sub>6</sub> H <sub>4</sub>	52	14	3i	81	99/1
10	$n - C_7 H_{15}$	p-Br-C <sub>6</sub> H <sub>4</sub>	56	48	3j	80	98/2
11	$n-C_8H_{17}$	E-1-hexenyl	54	20	3k	59	С
12	$n - C_8 H_{17}$	<i>p</i> -Me-C <sub>6</sub> H <sub>4</sub>	54	16	31	95	97/3 <sup>a</sup>
13	$n - C_8 H_{17}$	p-MeO-C <sub>6</sub> H <sub>4</sub>	54	16	3m	83	98/2 <sup>a</sup>
14	$n-C_8H_{17}$	Ph	58	18	3n	78	98/2
15	$n-C_8H_{17}$	p-Br-C <sub>6</sub> H <sub>4</sub>	56	18	30	46	99/14

<sup>*a*</sup> Diastereomer ratio; determined from 300 MHz <sup>1</sup>H NMR spectra of the crude products before separation. In all cases, except entries 4 and 7, the two isomers could not be separated via chromatography on silica gel using the conditions described in this paper. <sup>*b*</sup> The minor isomer could be isolated by chromatography on silica gel. <sup>*c*</sup> Cis isomer not observed. <sup>*d*</sup> The data were also confirmed by HPLC. <sup>*e*</sup> Determined from 400 MHz <sup>1</sup>H NMR spectrum.

epoxides for the synthesis of α-alkyl amino aldehydes and acids.<sup>9</sup> With our protocol for the highly diastereoselective synthesis of *trans*-2,3-disubstituted vinylic epoxides in hand, we utilized the highly enantioselective reduction of 1-undecyn-3-one to make optically active (*R*)-1-undecyn-3-ol,<sup>10</sup> which can be easily converted to the optically active (*R*)-1,2-dodecadien-4-ol (**1b**) (Scheme 3).<sup>7</sup> Its cyclization reaction with (1*E*)-hexenyl iodide, *p*-methylphenyl iodide, and *p*-methoxyphenyl iodide afforded (*R*,*R*)-**3k**, (*R*,*R*)-**3l**, and (*R*,*R*)-**3m**, respectively (Scheme 3).<sup>11</sup>

In conclusion, we have developed a mild, good-yielding, and highly diastereoselective methodology for the synthesis of *trans*-2,3-disubstituted vinylic oxiranes. Using the readily available chiral 2,3-allenols,<sup>12</sup> the corresponding vinylic oxiranes with high ee value can be synthesized.<sup>13–15</sup> Due to the wide synthetic importance of vinylic oxiranes, this methodology will show its utility in organic synthesis.





Acknowledgment. We thank the National Natural Science Foundation of China, Chinese Academy of Sciences, and Shanghai Municipal Committee of Science and Technology for financial support.

Supporting Information Available: Experimental details and <sup>1</sup>H NMR spectra of 3a-o (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

## JA9904888

<sup>(9)</sup> Jung, M. E.; D'Amico, D. C. J. Am. Chem. Soc. 1995, 117, 7379.
(10) Nishizawa, M.; Yamada, M.; Noyori, R. Tetrahedron Lett. 1981, 22, 247.

<sup>(11)</sup> wee was determined by HPLC with a CHIRALPAK AD column ( $\phi$  0.46 cm  $\times$  25 cm from Daicel Chemical Ind., Ltd.).

<sup>(12)</sup> According to Corey's method, chiral 2,3-dienols can be synthesized in one step via the reaction of aldehydes with chiral borane reagents. This makes our methodology highly attractive. Corey, E. J.; Yu, C.-M.; Lee, D.-H. J. Am. Chem. Soc. **1990**, *112*, 878.

<sup>(13) (</sup>a) For a seminal paper of the asymmetric Sharpless epoxidation reaction of allylic alcohols, see: Katsuki, T.; Sharpless, K. B. J. Am. Chem. Soc. **1980**, 102, 5974. (b) For kinetic resolution of racemic allylic alcohols, see: Martin, V. S.; Woodard, S. S.; Katsuki, T.; Yamada, Y.; Ikeda, M.; Sharpless, K. B. J. Am. Chem. Soc. **1981**, 103, 6237.

<sup>(14)</sup> For kinetic resolution of racemic terminal oxiranes, see: Tokunaga, M.; Larrow, J. F.; Kakiuchi, F.; Jacobsen, E. N. *Science* **1997**, 277, 936.

<sup>(15)</sup> For some of the most recent advances in the asymmetric epoxidation of simple alkenes, see: Zhang, W.; Loebach, J. L.; Wilson, S. R.; Jacobsen, E. N. J. Am. Chem. Soc. 1990, 112, 2801. Irie, R.; Noda, K.; Ito, Y.; Matsumoto, N.; Katsuki, K. Tetrahedron Lett. 1990, 31, 7345. Tu, Y.; Wang, Z.; Shi, Y. J. Am. Chem. Soc. 1996, 118, 9806. Yang, D.; Yip, Y.; Tang, M.; Wong, M.; Zheng, J.; Cheung, K. J. Am. Chem. Soc. 1996, 118, 491.